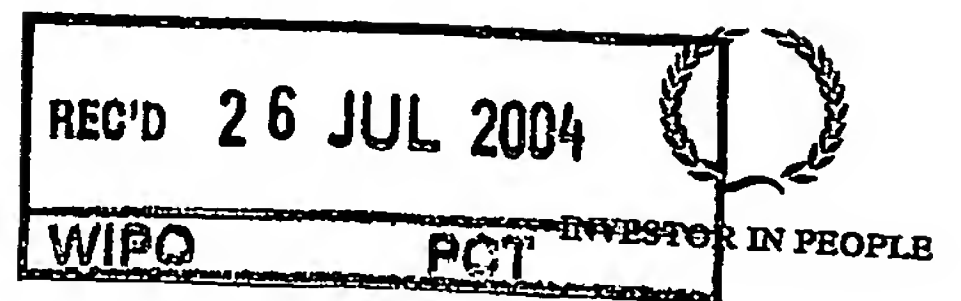




PCT/EP200 4 / 0 0 5 9 6 8



The Patent Office  
Concept House  
Cardiff Road  
Newport  
South Wales  
NP10 8QQ

BEST AVAILABLE COPY

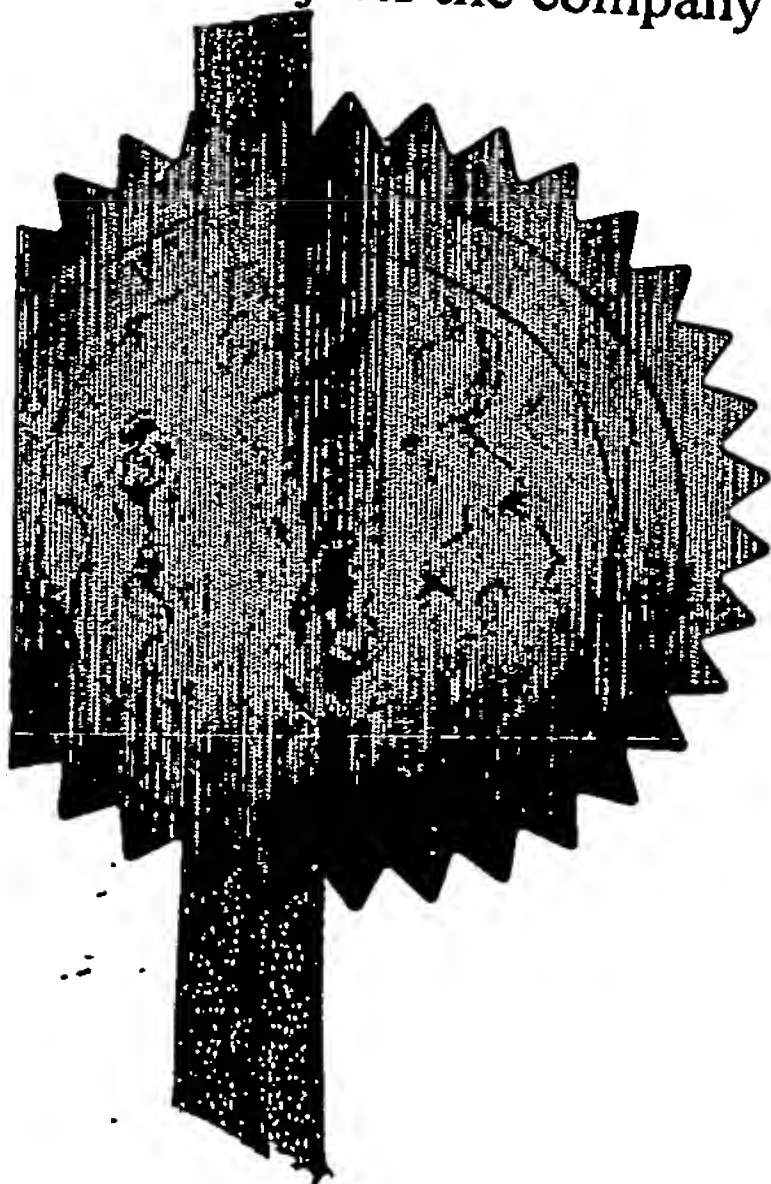
BEST AVAILABLE COPY

I, the undersigned, being an officer duly authorised in accordance with Section 74(1) and (4) of the Deregulation & Contracting Out Act 1994, to sign and issue certificates on behalf of the Comptroller-General, hereby certify that annexed hereto is a true copy of the documents as originally filed in connection with the patent application identified therein.

In accordance with the Patents (Companies Re-registration) Rules 1982, if a company named in this certificate and any accompanying documents has re-registered under the Companies Act 1980 with the same name as that with which it was registered immediately before re-registration save for the substitution as, or inclusion as, the last part of the name of the words "public limited company" or their equivalents in Welsh, references to the name of the company in this certificate and any accompanying documents shall be treated as references to the name with which it is so re-registered.

In accordance with the rules, the words "public limited company" may be replaced by p.l.c., plc, P.L.C. or PLC.

Re-registration under the Companies Act does not constitute a new legal entity but merely subjects the company to certain additional company law rules.



CERTIFIED COPY OF  
PRIORITY DOCUMENT

Signed

Dated 28 June 2004

**PRIORITY  
DOCUMENT**  
SUBMITTED OR TRANSMITTED IN  
COMPLIANCE WITH RULE 17.1(a) OR (b)

Patents Act 1977  
(Rule 16)

The  
Patent  
Office

# Request for grant of a patent

(See the notes on the back of this form. You can also get an explanatory leaflet from the Patent Office to help you fill in this form)

177  
JUN 12 2003  
E814726-1 CO3008  
POL/7/00 0.0040313668.6

The Patent Office

Cardiff Road  
Newport  
Gwent NP9 1RH

1. Your reference

PB - F3334

2. Patent application number  
(The Patent Office will fill in this part)

0313668.6

3. Full name, address and postcode of the or  
of each applicant (underline all surnames)

UNILEVER PLC  
Unilever House,  
Blackfriars,  
London, EC4P 4BQ

12 JUN 2003

Patents ADP number (if you know it)

1628002

If the applicant is a corporate body, give the country/state of its incorporation

Great Britain

4. Title of the invention

METHOD OF CONTAINING PEAS

5. Name of your agent (if you have one)

Lloyd Wise

"Address for service" in the United Kingdom to which all correspondence should be sent (including the postcode)

Commonwealth House,  
1 - 19 New Oxford Street,  
London, WC1A 1LW

Patents ADP number (if you know it)

117001

If you are declaring priority from one or more earlier patent applications, give the country

Country

Priority application  
number  
(if you know it)

Date of filing  
(day / month / year)

and the date of filing of the or of each of these earlier applications (and if you know it) the or each application number

If this application is divided or otherwise derived from an earlier UK application, give the number and the filing date of the earlier application

Number of earlier application

Date of filing  
(day / month / year)

Is a statement of inventorship and of right to grant of a patent required in support of this request (Answer 'Yes' if:

Yes

- a) any applicant named in part 3 is not an inventor, or
- b) there is an inventor who is not named as an applicant, or
- c) any named applicant is a corporate body. See note (d))

**Patents Form 1/77**

9. Enter the number of sheets for any of the following items you are filing with this form.

Do not count copies of the same document

Continuation sheets of this form

Description 14

Claim(s) 3

Abstract

Drawing(s) 6 only

OF

10. If you are also filing any of the following, state how many against each item.

Priority documents

Translation of priority documents

Statement of inventorship and right to grant of a patent (*Patents Form 7/77*)

Request for preliminary examination x  
and search (*Patents Form 9/77*)

Request for substantive examination  
(*Patents Form 10/77*)

Any other documents  
(please specify)

11. I/We request the grant of a patent on the basis of this application.

Signature

12th June, 2003

LLOYD WISE

12. Name and daytime telephone number of person to contact in the United Kingdom PAUL BOWMAN  
020 571 6200

**Warning**

After an application for a patent has been filed, the Comptroller of the Patent Office will consider whether publication or communication of the invention should be prohibited or restricted under Section 22 of the Patents Act 1977. You will be informed if it is necessary to prohibit or restrict your invention in this way. Furthermore, if you live in the United Kingdom, Section 23 of the Patents Act 1977 stops you from applying for a patent abroad without first getting written permission from the Patent Office unless an application has been filed at least 6 weeks beforehand in the United Kingdom for a patent for the same invention and either no direction prohibiting publication or communication has been given, or any such direction has been revoked.

**Notes**

- a) If you need help to fill in this form or you have any questions, please contact the Patent Office on 0645 500505.
- b) Write your answers in capital letters using black ink or you may type them.
- c) If there is not enough space for all relevant details on any part of this form, please continue on a separate sheet of paper and write "see continuation sheet" in the relevant part(s). Any continuation sheet should be attached to this form.
- d) If you have answered 'Yes' Patents Form 7/77 will need to be filed.
- e) Once you have filled in the form you must remember to sign and date it.
- f) For details of the fee and ways to pay, please contact the Patent Office.

The oxygen concentration may be maintained by establishing a continuous gas flow, for example airflow, relative to the peas. The gas flow may be established by using pumps, fans or the like to blow gas through the fresh pea product. It has been found that a gas flow rate of between  $1000\text{Lmin}^{-1}$  and  $14000\text{Lmin}^{-1}$  flowing through each square metre of the load in a plane generally perpendicular to the air flow, for example flowing upwardly through each square metre of the base of the container supporting the load, can maintain the oxygen and carbon dioxide concentration at appropriate levels. Preferably, the flow rate is between  $1280\text{Lmin}^{-1}\text{m}^{-2}$  and  $12800\text{Lmin}^{-1}\text{m}^{-2}$ . More preferably, the flow rate is between  $1280\text{Lmin}^{-1}\text{m}^{-2}$  and  $6000\text{Lmin}^{-1}\text{m}^{-2}$ .

Alternatively, the peas may be moved relative to the gas, for example by a tumbling action. In one embodiment, the movement of the container, for example by movement of the lorry, causes air to be distributed or circulated around the peas. This has the advantage that no separate pump or fan is required.

According to one embodiment, the container can be provided with a conduit having an inlet in communication with the outside atmosphere and an outlet in communication with the interior of the container. When the container is mounted on a vehicle, the inlet preferably faces the forward direction of travel of the vehicle, thus causing air to flow from outside the container to the interior of the container when the vehicle is moving.

This embodiment of the invention is particularly advantageous since no moving parts are required, it being the movement of the container caused by the movement of the vehicle that maintains the oxygen concentration. Particularly good results may be achieved if the outlet of the conduit is positioned near the bottom of the container, so as to cause air to flow in an upward direction through the pea load. This ensures that entire the load is aerated.

It has been found that a cross sectional area (C) of between approximately  $0.072\text{m}^2$  and  $0.13\text{m}^2$  for the inlet of the conduit can sufficiently aerate the pea load. A conduit having a cross sectional area between these values takes in enough air, for vehicle speeds between  $4.7\text{ms}^{-1}$  (10mph) and  $26.3\text{ms}^{-1}$  (56mph), to provide a flow rate of



between  $1280\text{Lmin}^{-1}\text{m}^{-2}$  and  $12800\text{Lmin}^{-1}\text{m}^{-2}$ . It has been found that only about 50% of the air flowing over the vehicle towards the conduit actually enters the conduit. This has been taken into consideration when calculating the above figures. A more general relationship has been derived for the relationship between the area of the tank-floor (F) and the duct-inlet cross sectional area (C). To meet minimum air-flow requirement, the cross sectional area must satisfy  $C_{\min} = 0.00454 \times F$ , for maximum air-flow the relationship is  $C_{\max} = 0.00811 \times F$ .

The container may comprise a main part to receive the peas and a cavity to receive the outlet of the conduit. A partition may separate the cavity and the main part and may allow gases such as oxygen and air to flow from the cavity into the main part. In this way, an even distribution of airflow or oxygen flow through the product can be obtained. The cavity may be formed on one side of the container, but preferably it is formed at the bottom of the container.

The partition may be formed from a food grade, plastic material, which, advantageously, is hard wearing. It is beneficial if the partition can be easily cleaned. The partition may be porous, and preferably is formed with pores in the form of holes which are distributed over the partition to allow an even airflow across the surface of the material without creating back pressures thereby allowing free airflow through the partition. The partition may comprise a perforated sheet of metal or plastics material. Alternatively, the partition may be formed from another hydrophobic material. This can also reduce the likelihood of liquid clogging the holes.

A preferred material is VYON<sup>®</sup>, a porous plastic material produced by Porvair. VYON<sup>®</sup> can have various thicknesses and pore sizes. For example, the thickness can be from  $7.5 \times 10^{-4}\text{m}$  to  $100 \times 10^{-4}\text{m}$ . The pores may have diameters between  $2 \times 10^{-6}\text{m}$  and  $300 \times 10^{-6}\text{m}$ . Preferably, the pores have an average diameter of  $98 \times 10^{-6}\text{m}$ . VYON<sup>®</sup> can be formed from different materials, such as high density polyethylene, ultra high molecular weight polyethylene or polypropylene.

The container may have an opening, for example in an upper surface, which can be provided with a removable cover. The cover must allow the flow of gas between the container and the outside atmosphere. In this way, the build up of pressure within

the container is avoided. The cover may be porous or perforated or separated slightly from the container. Alternatively, the container may be formed with one or more openings which are not under the cover.

5 In a further embodiment, the container can be agitated, or can be provided with a part that may be agitated, to generate an airflow relative to the peas. For example, the container or a part thereof may rotate along a generally horizontal or slightly inclined axis so as to cause the peas to tumble. The container may have ribs or baffles formed on its interior surface which lift the peas as the container rotates. This can result in improved aeration of the peas. This embodiment may operate in much  
10 the same way as a cement mixer. This method of aerating the peas has the advantage that the level of oxygen in ambient air is maintained around the peas.

The container may be in fluid communication with the outside atmosphere, for example by way of a conduit or an opening in the container. Optionally, the opening  
15 may be guarded by a removable cover, which allows the flow of gas between the container and the outside atmosphere, and prevents loss of produce and entry of foreign material.

In all embodiments, the oxygen or air to be distributed throughout the pea load may first be humidified. Methods of humidifying oxygen and air known to those skilled in the art may be used, for example by passing the gas through water or by way of an  
20 aerosol device. This prevents the flow of oxygen or air having a drying effect on the peas, which can reduce the flavour, texture and quality in general of the peas.

A filter may be used to filter the oxygen or air before it is distributed throughout the pea load. This is particularly advantageous where air flows from outside the container into the container due to the movement of the vehicle, since impurities such  
25 as dust and insects need to be removed.

The concentration by volume of oxygen in the air throughout the pea load is maintained at or above 15%. It is further preferred that the concentration by volume of oxygen throughout the pea load is maintained at or above 18%. It is particularly preferred that the concentration by volume of oxygen throughout the pea load is  
30 maintained at or above ambient level, i.e. approximately 20.9% by volume. These

levels can readily be achieved using ambient air, although oxygenated air may be employed.

5 The concentration by volume of carbon dioxide throughout the pea load is maintained at or below 1% by volume. It is further preferred that the concentration by volume of carbon dioxide throughout the pea load is maintained at or below 0.25%. It is particularly preferred that the concentration by volume of carbon dioxide throughout the pea load is maintained at or below ambient level, i.e. approximately 0.03%. The defined levels of carbon dioxide can readily be achieved by aeration.

10 Embodiments of the present invention will now be described by way of a number of examples, with reference to the following Figures, in which

Figure 1a is a graph showing the oxygen level surrounding the peas in a pea load transported using the known method;

Figure 1b is a graph showing the carbon dioxide level surrounding the peas in a pea load transported using the known method;

15 Figure 2 is a schematic diagram of a container suitable for use in the present invention;

Figure 3a is a graph showing the oxygen level in the container of Figure 2 compared with the oxygen level in a similar container which is not aerated;

20 Figure 3b is a graph showing the carbon dioxide level in the container of Figure 2 compared with the carbon dioxide level in a similar container which is not aerated;

Figure 4 is a schematic diagram of another container for use in the present invention, which is mounted on a vehicle;

Figure 5a is a graph showing the final oxygen level in various pea loads;

Figure 5b is a graph showing the final carbon dioxide level in various pea loads;

25 Figure 6 is a schematic diagram of another container for use in the present invention.

Figure 7 is a graph showing ethanol content after processing

As a first, comparative example, the oxygen and carbon dioxide levels of two commercial pea loads were monitored from the field to the factory. Peas of the Waverex variety were harvested and then introduced into a known pea transporter. Approximately 2500kg of peas were placed in the transporter. A gas sampling lance, which is described in more detail below, was inserted and positioned near the bottom of the load, at a depth of 60cm. A 50ml gas sample was then taken and a timer was started. The pea load was then transported to a factory and further 50ml gas samples were taken at various times during the journey, the final one being at the factory. The gas samples were then analysed using a Gaspac 2 Analyser (model ZR869, available from Systech Instruments, Thame, England) to determine the oxygen and carbon dioxide levels.

Another known pea container was filled with peas of the Barle variety. The peas were transported and gas samples were taken in the same way as described above. However, in this case, two gas sampling lances were used, one at a depth of 90cm and one at a depth of 5cm. The Barle load was deeper than the Waverex load.

The gas sampling lance was used to take samples of gas from within the pea load for later analysis. The lance comprised a rigid plastic tube, 180cm in length and 35.7mm in diameter, and had a cone-shaped plug attached to the distal end. A tube, 300cm in length and 1.57mm in internal diameter was fed through the outer tube to the cone plug. A 560cm extension portion terminating in the cab of the vehicle, having a switcher, a pump and a 50ml sampling syringe associated therewith, was added to enable gas samples to be taken from the container during transit. The gas samples were analysed for oxygen and carbon dioxide content after the journey, using the Gaspac 2 Analyser.

The results of the above two experiments are shown in Figures 1a and 1b. Figure 1a shows the variation in oxygen level with journey time and Figure 1b shows the variation in carbon dioxide with journey time. In each Figure, concentration of oxygen or carbon dioxide respectively is plotted on the y-axis, and time in minutes is plotted along the x-axis.



Data series A represents measurements of oxygen and carbon dioxide concentration respectively at the depth of approximately 60cm, for the pea variety "Waverex";

Data series B represents measurements of oxygen and carbon dioxide concentration respectively at the depth of approximately 90cm, for the pea variety "Barle";

- 5 Data series C represents measurements of oxygen and carbon dioxide concentration respectively at the depth of approximately 5cm, for the pea variety "Barle".

10 It can be seen from Figure 1a that the initial oxygen concentration for all three data series was generally equal to that in ambient air. However, the concentration near the bottom of the load decreases rapidly with time for both the Waverex and Barle pea varieties. After 60 minutes, the oxygen concentration was 85% less than the initial concentration. It can be seen that the results for the two varieties are very similar.

15 The concentration near the top of the Barle load also decreased with time. After 60 minutes, the oxygen concentration had dropped by 32%. It appears that aeration naturally occurs to a small extent in the top level of the load.

It can be seen from Figure 1b that the concentration of carbon dioxide was found to increase significantly with time. Near the bottom of the pea load, the carbon dioxide concentration after 60 minutes was 110 times the initial concentration.

20 A second example employed the apparatus shown in Figure 2. This apparatus is suitable for use in the present invention on a small, experimental scale. The container 1 is an upright tube capable of containing peas to a similar depth as a normal load. A plastic container having a height of 100cm and an internal diameter of 100mm was used. The container has a cavity 6 at its bottom, and a partition 7 separating the cavity 6 and a main part 2 of the container and upon which the peas are loaded. The partition 7 is formed from a grid, mesh or porous material so as to allow the flow of gas from the cavity 6 to the main part 2 of the container. A preferred material is VYON<sup>®</sup>, a porous plastic material produced by Porvair. In this apparatus, VYON<sup>®</sup> HP was used, having a thickness of  $2.5 \pm 0.25$  mm and an average pore size of  $98 \times 10^{-6}$  m.

The container 1 has an opening 8 adjacent or near the partition 7 for a syringe or other apparatus to be inserted. This enables samples of gas to be taken from near the partition 7, enabling the determination of the level of oxygen, carbon dioxide or other components of the gas. A 50ml syringe was used to take gas samples, and the samples were analysed using a Gaspac 2 analyser.

A line 9 from a compressed air bottle (not shown) has an outlet in the cavity 6 and allows air to pass from a bottle to the cavity 6. The air passes through the porous partition 7 and aerates the peas from below, maintaining the level of oxygen around the peas. Oxygen or oxygenated air could also be used to maintain the oxygen concentration throughout the pea load.

A control container which had an identical structure, but which was not connected to a gas supply was used as a control.

In a first part of this experiment, both containers were filled with 10kg of freshly harvested peas of the Bikini variety to a depth of 100cm, and a sample of gas was immediately taken from each. One container was aerated by an airflow rate of approximately 5 litres per minute. A gas sample was then taken from each container after 120 minutes.

In a second part of this experiment, both containers were filled with 10kg of freshly harvested peas of the Novella variety to a depth of 100cm, and a sample of gas was immediately taken from each. One container was aerated by an airflow rate of approximately 5 litres per minute. A gas sample was then taken from each container after 75 minutes.

The results obtained for these experiments are shown in Figures 3a and 3b. Oxygen levels are shown in Figure 3a and carbon dioxide levels are shown in Figure 3b. The results for the Bikini variety are labelled D and the results for the Novella variety are labelled E.

Column 1 shows the oxygen or carbon dioxide levels respectively in the gas samples taken at the start;

Column 2 shows the oxygen or carbon dioxide levels respectively in the control gas samples taken at the end of the experiment;

Column 3 shows the oxygen or carbon dioxide levels respectively in the aerated gas samples taken at the end of the experiment.

- 5 It can be seen that, at the end of the experiment, the level of oxygen near the bottom of the aerated container is generally the same as the initial level. In comparison, the level of oxygen near the bottom of the control container is much lower at the end of the experiment. Similarly, the results show that the level of carbon dioxide in the container can be greatly reduced by aerating the container.
- 10 A third example illustrates how the present invention can be used on a larger scale. The apparatus shown in Figure 4 comprises a container 1 which has a 50cm x 50cm square base and a height of 75cm. The container was formed from plastic, although other rigid materials such as metal would also be suitable. The container 1 is mounted on a vehicle 11 to allow peas to be transported therein from the field to the
- 15 factory. The container 1 has a cavity 6 at its bottom and a partition 7 upon which the peas are loaded. The peas are therefore contained in the main part 2 of the container 1.

The partition 7 is formed from a grid, mesh or porous material so as to allow the flow of gas from the cavity 6 to the main part 2. A preferred material is VYON<sup>®</sup>, a porous plastic material produced by Porvair. In this apparatus, VYON<sup>®</sup> HP was used, having

20 a thickness of  $2.5 \pm 0.25$  mm and an average pore size of  $98 \times 10^{-6}$  m.

The container 1 has a cover 12 that is preferably removable. The cover 12 is positioned so as to allow air to escape the container 1, thus avoiding a build up of pressure, and also to prevent the accumulation of rainwater in the container.

25 Accordingly, the cover 12 can be positioned slightly above the container 1, or the top of the container 1 can be formed with one or more openings. The cover 12 may be made from a rigid material such as plastic or metal, or may be a tarpaulin.

A conduit 3 has an inlet 5 facing the forward direction of travel of the vehicle, and an outlet 4 inside the cavity 6. The forward movement of the vehicle 11 therefore

causes air to flow through the conduit 3, into the cavity 6 and through the porous partition 7 into the pea load.

The conduit 3 optionally has a filter 14 to filter dust particles, insects or other impurities from the air. Additionally, the conduit 3 can have an anemometer and a device, such as an aerosol device, to humidify the incoming air.

An opening, such as a hatch 15, may be formed in a wall of the container, adjacent or near the partition 7, to allow peas to be removed from the container. Further openings 16 are formed in the wall of the container to allow one or more sampling tubes to be inserted into the container to draw gas samples. The tube is preferably made of plastic and has a length of 25cm and an internal diameter of 6mm.

Two containers as shown in Figure 4 were filled with approximately 80kg of freshly harvested peas to a depth of 70cm. Covers were positioned on the containers and gas samples were taken from both. The peas in one container were then aerated. Both containers were transported to the factory. Further gas samples were taken from each container during the journey, the final samples being taken on arrival at the factory. The ambient temperature was noted each time a gas sample was taken.

To provide a comparison, two containers as shown in Figure 2 were filled with peas. One container was aerated and the other was not. Gas samples were taken from each container at various times, and the ambient temperature was recorded simultaneously.

The variation of oxygen and carbon dioxide levels in this example are shown in Figures 5a and 5b respectively. The oxygen and carbon dioxide levels are shown along the y-axis, and the duration of the journey (in minutes) multiplied by the ambient temperature (in degrees centigrade) is shown along the x-axis.

25 Data series F refers to readings taken from a control container according to Figure 4;

Data series G refers to readings taken from an aerated container according to Figure 4;

Data series H refers to readings taken from a control container according to Figure 2;



Data series J refers to readings taken from an aerated container according to Figure 2.

5 It can be seen from Figures 5a and 5b that, by aerating peas contained in the apparatus shown in Figure 2 or Figure 4 a high, approximately ambient level of oxygen and a reduced level of carbon dioxide can be maintained in the peas.

It has been found that an increased temperature in the pea load contributes to oxygen depletion within the load. Accordingly, passing air or oxygen through the pea load may additionally help maintain the quality of the peas by cooling the pea load.

10 It is to be understood that the apparatus shown in Figure 4 could be adjusted to have different dimensions. A structurally similar but larger container could be used to transport commercial loads.

15 Figure 6 shows a further embodiment of the present invention, in which the peas are aerated by tumbling them in a rotating container 1. The container rotates about the slightly inclined axis, Z, although a generally horizontal axis could also be used. No mechanism to cause air to flow through the pea load is required, since tumbling the peas aerates the load sufficiently.

20 According to a fourth example, the apparatus of Figure 2 was used to determine and compare the ethanol content of aerated peas with the ethanol content of non-aerated peas. Ethanol content is a typical product of anaerobic oxidation (eg in brewing), and the level will indicate the effectiveness of the aeration treatment in maintaining the condition of the harvested peas. Experiments were conducted in which one container of peas was aerated and another, control container, was not.

In Experiment K, Bikini peas were stored for 120 minutes at a temperature of 21°C.

25 In Experiment L, Novella peas were stored for 120 minutes at a temperature of 28.5°C.

In Experiment M, Novella peas were stored for 75 minutes at a temperature of 15.8°C.

At the end of each experiment, approximately 1.5kg of peas was taken from the bottom of the both the aerated container and the control container. These samples were washed, blanched in Netlon bags and frozen (-35°C). 10g of each sample were placed in 10ml of cold water (4°C) and macerated using an Ultra Turrax (maximum speed). The slurry from each sample was then centrifuged for 20 minutes at 12,000g and the supernatant collected as an extract. A YSI 2700 Select biochemistry analyser (black electrode, standard recommended setting) (available from Analytical Technologies, Farnborough, England) was used to assess the level of ethanol in each extract.

The results of this example are shown in Figure 7. Ethanol content in parts per million is indicated along the y-axis. The results for each Experiment are labelled K, L and M respectively. Column 1 shows the ethanol level in the control peas and Column 2 shows the ethanol level in aerated peas.

As can be seen, the ethanol content of the non-aerated samples is greater than that of the aerated samples. It is believed that the ethanol content in non-aerated peas is actually much greater than that in aerated peas prior to blanching, but that the ethanol is washed out during blanching.

It can be seen from the results that storage time and temperature has an effect on the amount of ethanol produced. The Novella peas of Experiment L produced more ethanol than the peas in the other two experiments, since they were stored for longer at a higher temperature. Similarly, the peas in Experiment M produced less ethanol, since the storage time was relatively short and the temperature was much lower.

Additionally, the difference in ethanol content between the non-aerated sample and the aerated sample is greater in Experiment L than in Experiment K, which is in turn greater than in Experiment M.

These experiments demonstrate that aeration effectively reduces anaerobic fermentation and therefore ethanol production is suppressed. Also, aeration may have a cooling effect that lowers the temperature of the peas, due to evaporative cooling effects that also suppress fermentation.

Two trained pea tasters tasted peas from each of the three Experiments, to determine whether the treatment improved quality of the produce post harvest. In respect of Experiment K, the aerated peas were preferred. In respect of Experiment L, the aerated peas were preferred and a very clear difference was detected. In respect of Experiment M, no overall preference was found.

It is understandable that there was no overall preference in Experiment M. As mentioned above, the conditions in which the peas were stored, namely low temperature and for a short period of time, meant that quality loss was not as rapid compared with conditions of higher temperature or longer storage periods.

10 In any of the above embodiments, the air or oxygen used to maintain the oxygen levels can be humidified so as to avoid a drying effect.

## CLAIMS

- 1     A method of containing peas comprising the steps of providing a container  
containing a pea load, maintaining the concentration of oxygen at or above 15% by  
volume throughout the container and maintaining the concentration of carbon dioxide  
5     at or below 1% by volume throughout the container.
- 2     A method of containing peas according to Claim 1, wherein the container is  
mounted on a vehicle.
- 3     A method of containing peas according to Claims 1 or 2, wherein a conduit is  
provided having an inlet in communication with the outside atmosphere and an outlet  
10     in communication with the interior of the container.
- 4     A method of containing peas according to Claims 3 when dependent on Claim  
2, wherein the inlet faces the forward direction of travel of the vehicle.
- 5     A method of containing peas according to Claim 3 or Claim 4, wherein the  
outlet of the conduit is positioned near the bottom of the container.
- 15     6     A method of containing peas according to any of Claims 3 to 5, wherein the  
container comprises a main part to receive the peas and a cavity to receive the outlet  
of the conduit, the cavity being formed at the bottom of the container and being  
separated from the main part by a gas permeable partition.
- 7     A method of containing peas according to Claims 1 or 2, wherein at least a  
20     part of the container rotates about an inclined axis.
- 8     A method of containing peas according to any previous claim, wherein a fan  
establishes a gas flow through the peas.
- 9     A method of containing peas according to any previous claim, wherein a gas  
flow rate, flowing through each square metre of the pea load in a plane generally  
25     perpendicular to the air flow, is between 1000Lmin<sup>-1</sup> and 14000Lmin<sup>-1</sup>.



- 10     A method of containing peas according to any previous claim, wherein the container has an opening which is provided with a removable cover and which allows the flow of gas between the container and the outside atmosphere.
- 5     11     A method of containing peas according to any previous claim, wherein the concentration of oxygen throughout the pea load is maintained at or above atmospheric air levels.
- 12     A method of containing peas according to any previous claim, wherein the concentration of carbon dioxide throughout the pea load is maintained at or below atmospheric air levels.
- 10     13     A container for containing peas provided with means for establishing a gas flow through the container to maintain the oxygen concentration at or above 15% by volume throughout the container and the carbon dioxide concentration at or below 1% by volume throughout the container.
- 14     A container for containing peas according to Claim 13, mounted on a vehicle.
- 15     15     A container for containing peas according to Claims 13 or 14, the container comprising a main part, in which in use the peas are received, and a cavity, the main part and the cavity being separated by a gas permeable partition, the container being provided a conduit having an inlet and an outlet, the outlet being located in the cavity and the inlet being located outside the container.
- 20     16     A container for containing peas according to Claim 15 when dependent on Claim 14, wherein the inlet of the conduit faces the forward direction of travel of the vehicle.
- 17     A container according to Claims 13 or 14, containing peas, wherein at least a part of the container rotates about a slightly inclined axis.
- 25     18     A method of processing peas including the steps of harvesting peas, introducing the peas into a container according to any of Claims 13 to 17, transporting the peas to a processing station, blanching the peas and freezing the peas.

F3334

17

19 A method of processing peas according to Claim 18, wherein the peas are then packaged.

---

Figure 1a

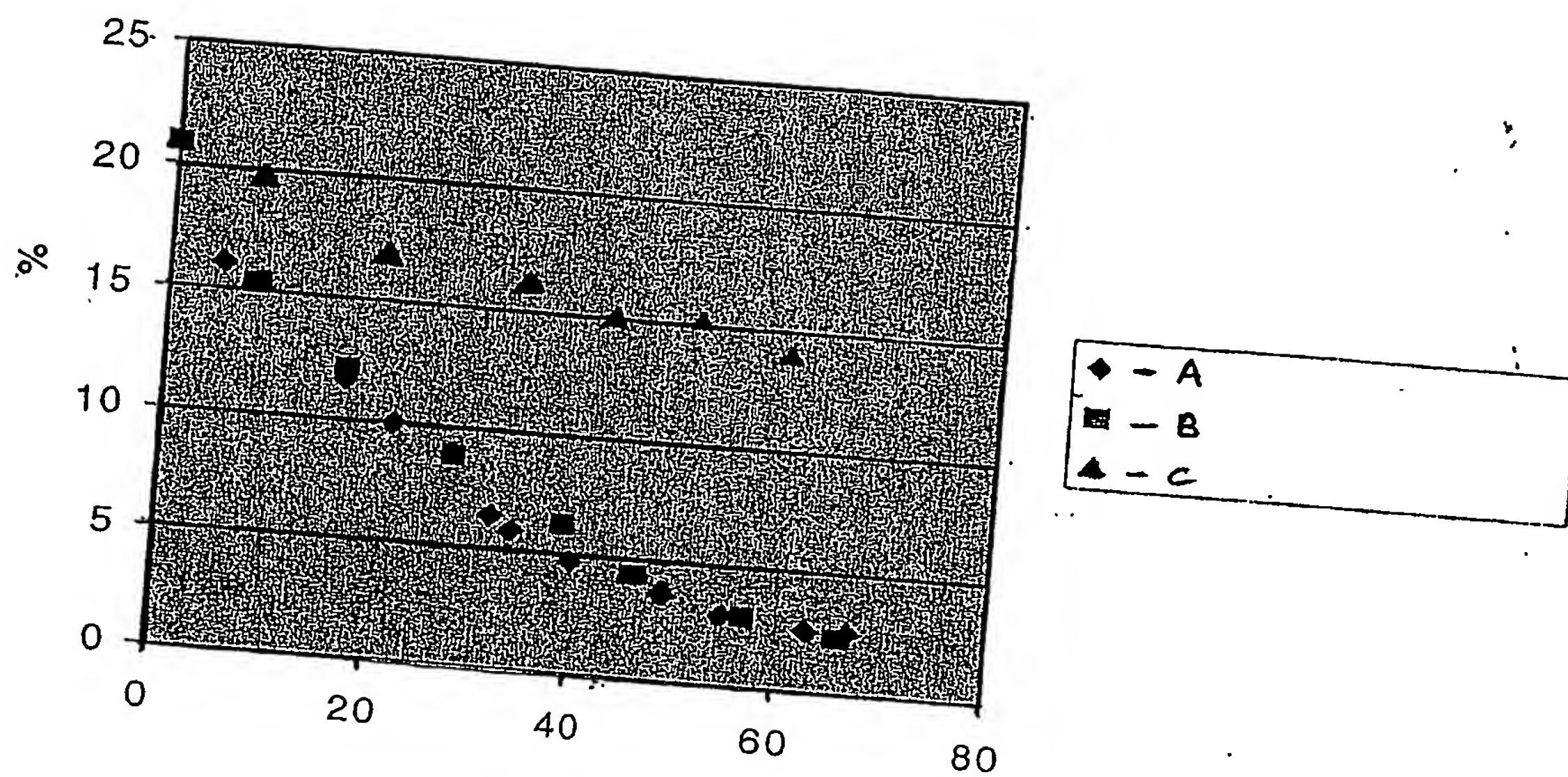


Figure 1b

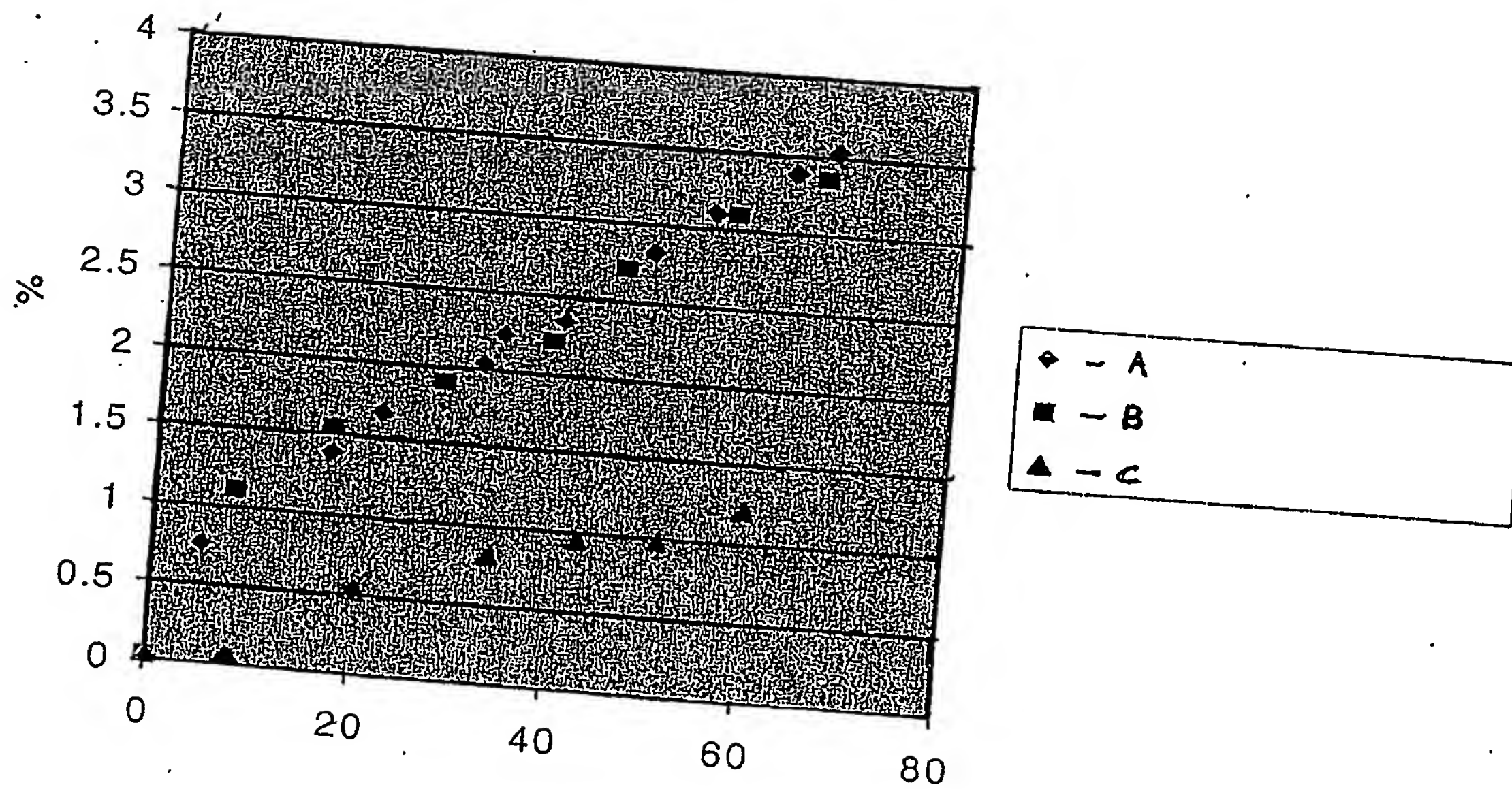
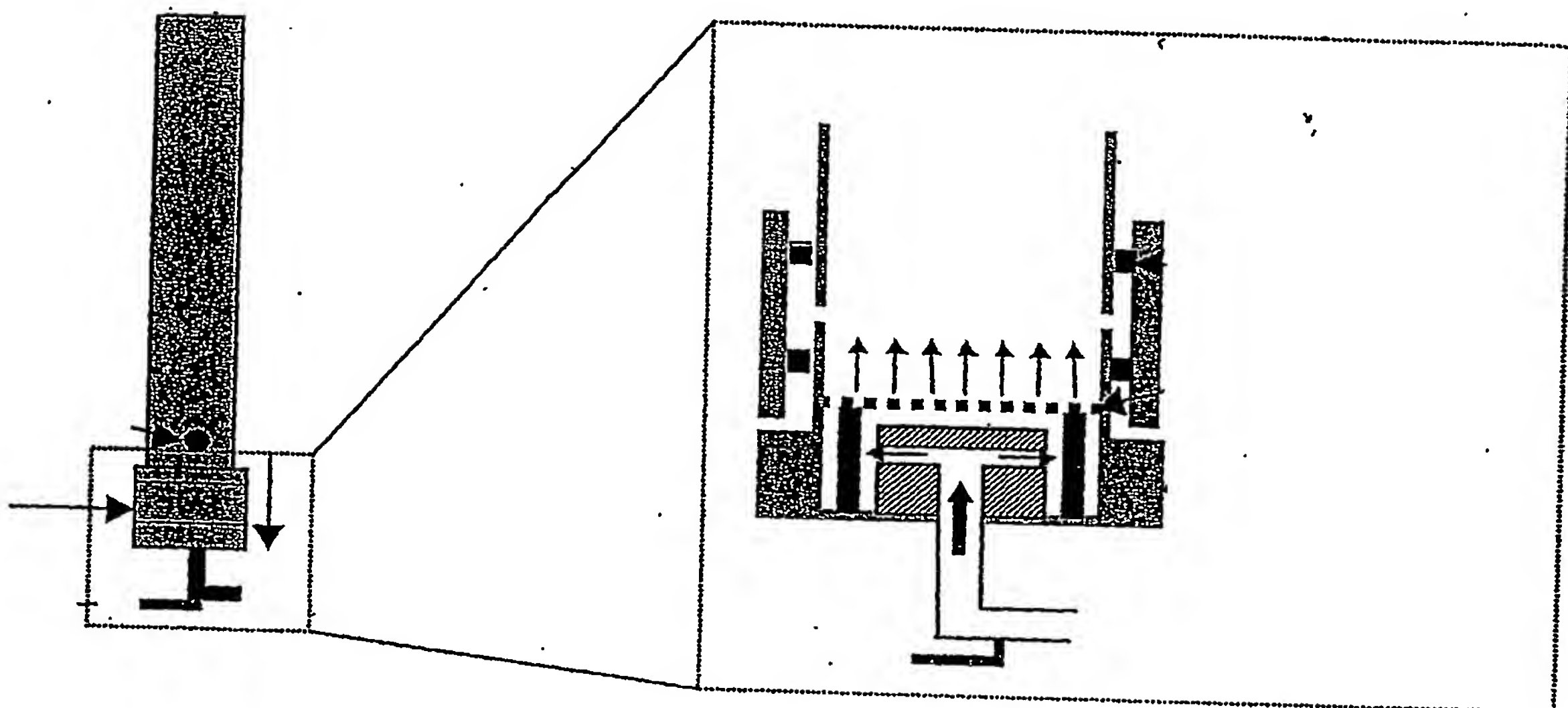


Figure 2





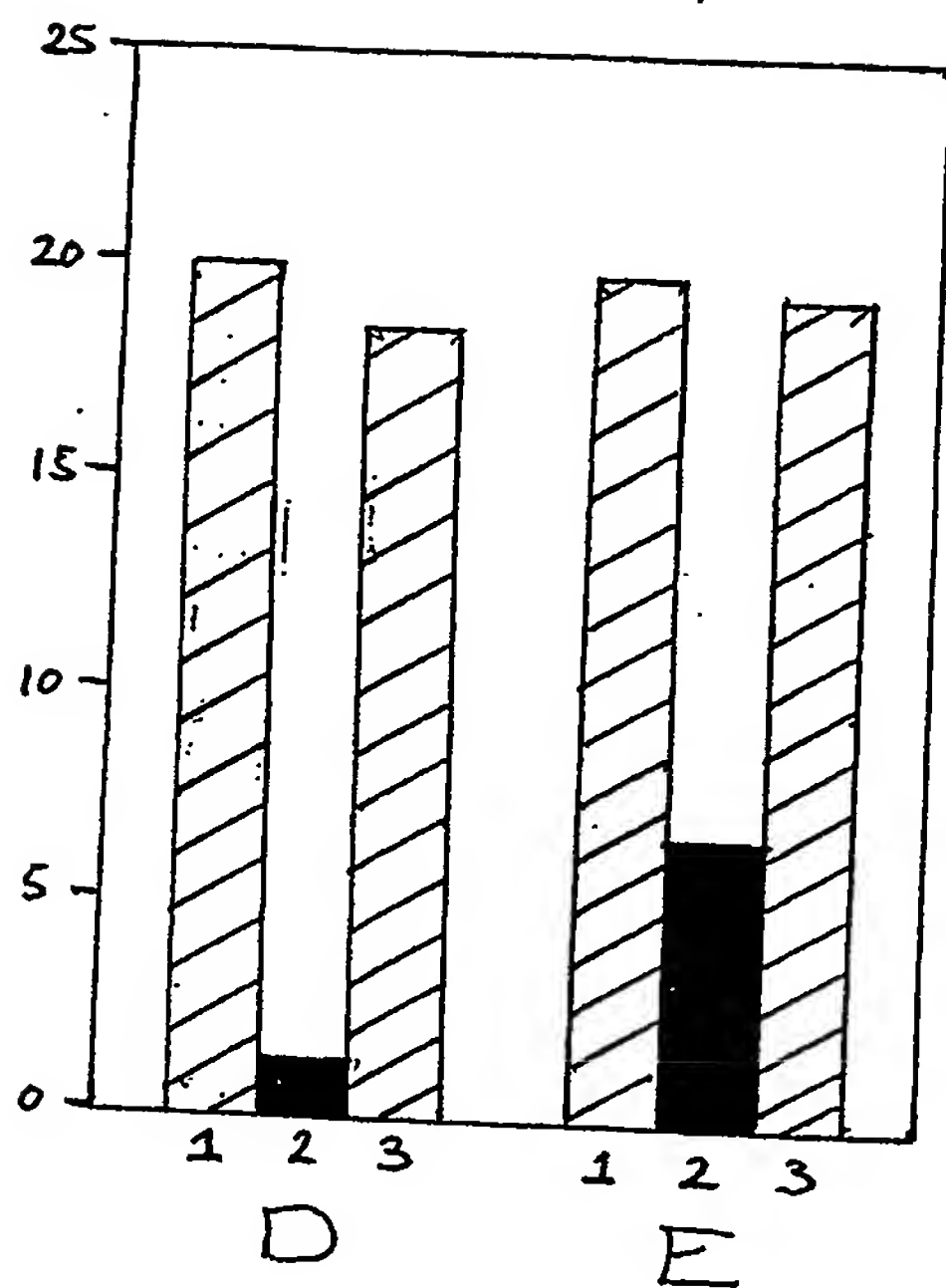


Figure 3a

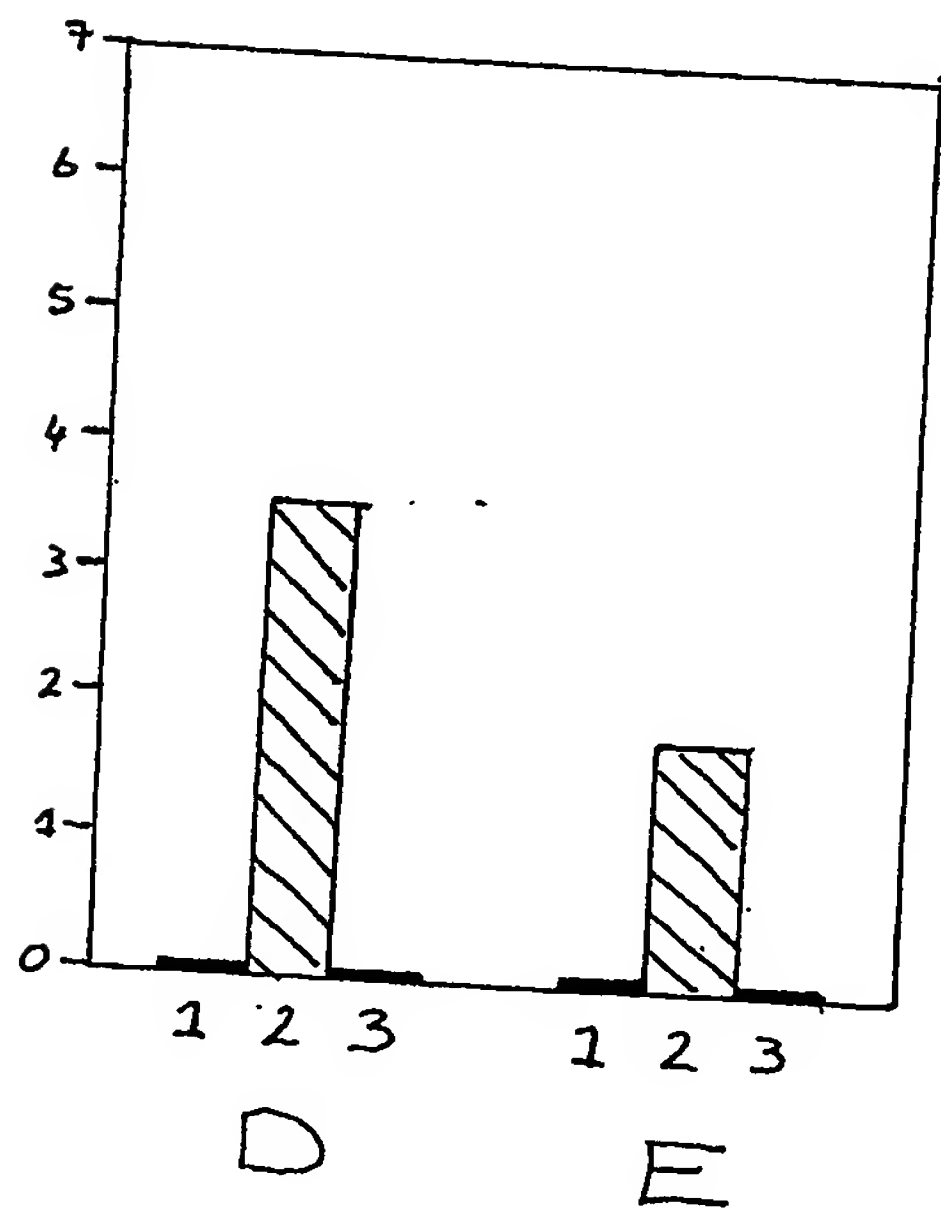


Figure 3b

Figure 4

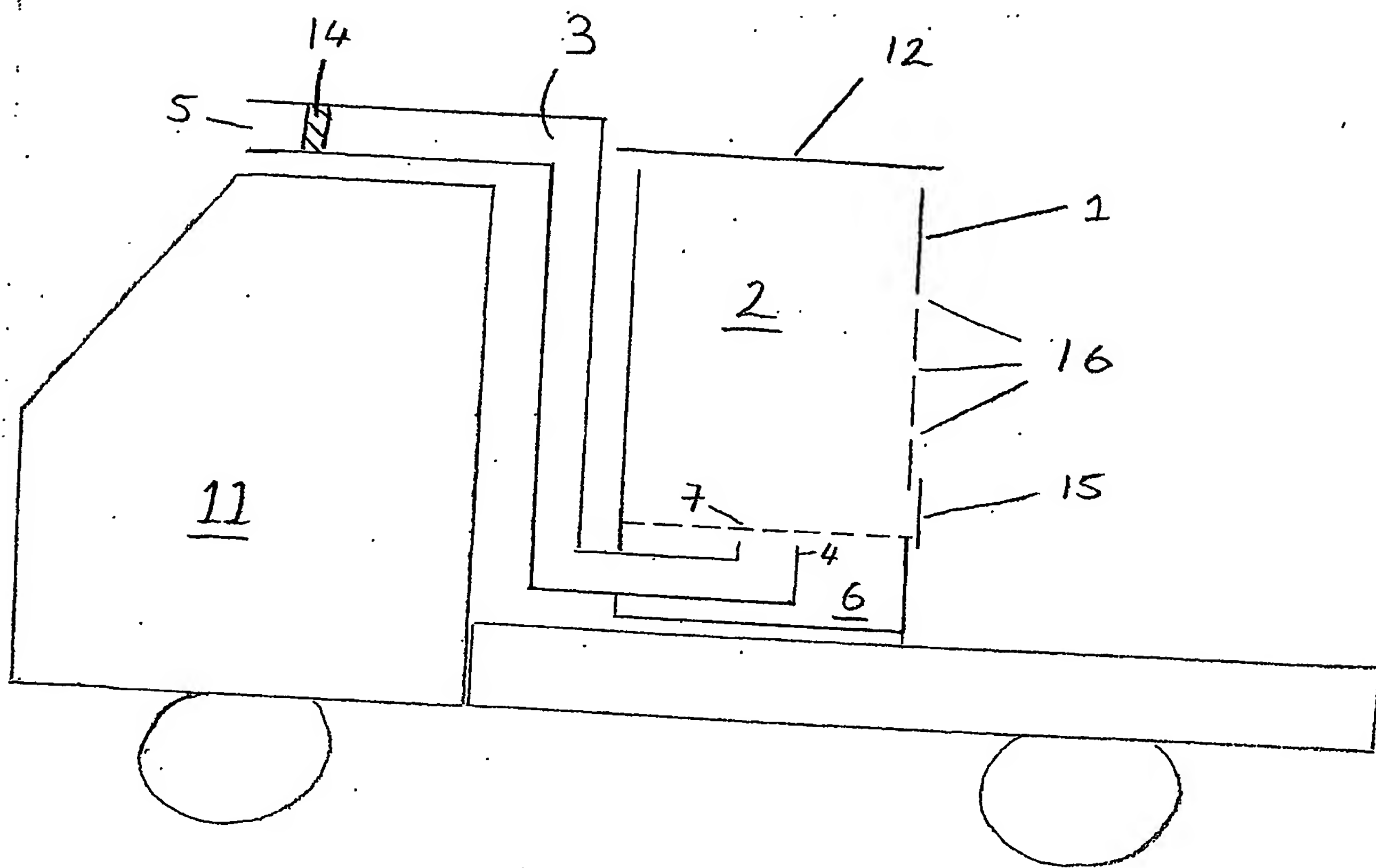


Figure 5a

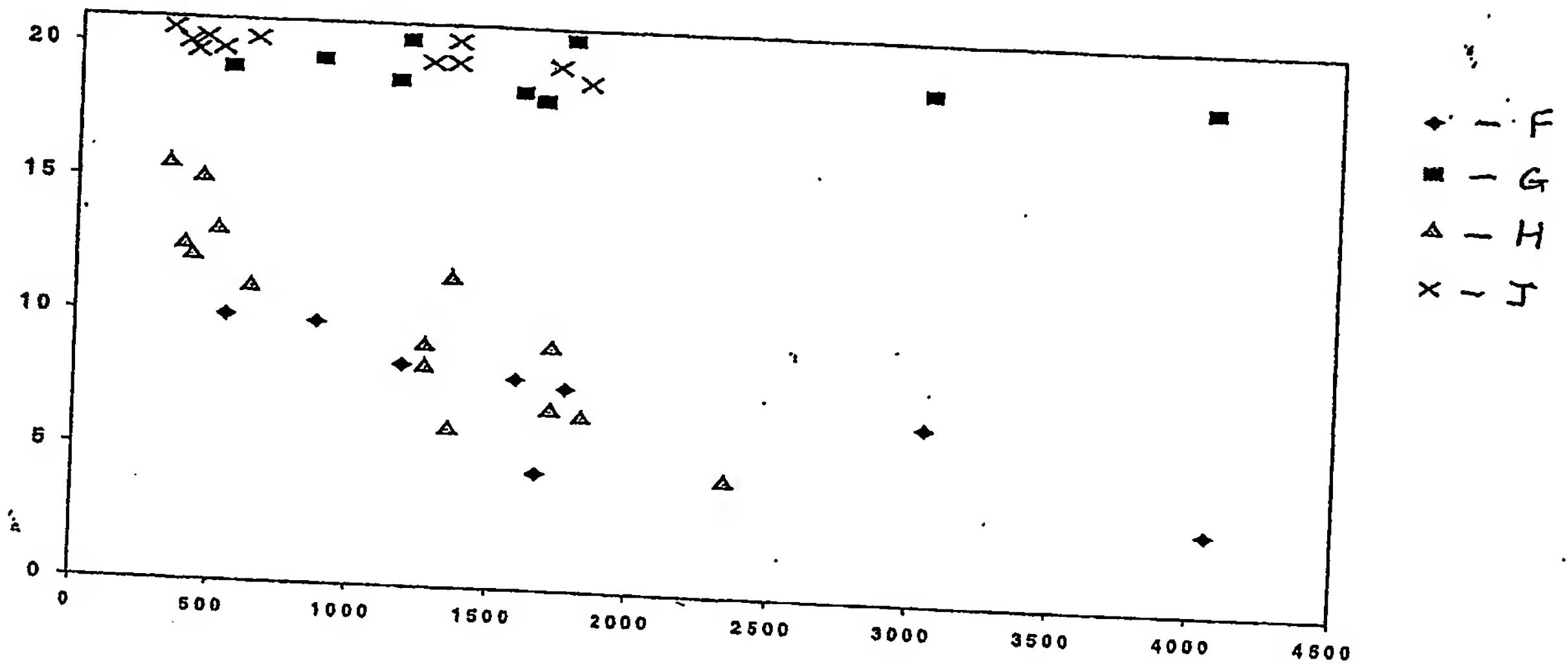


Figure 5b

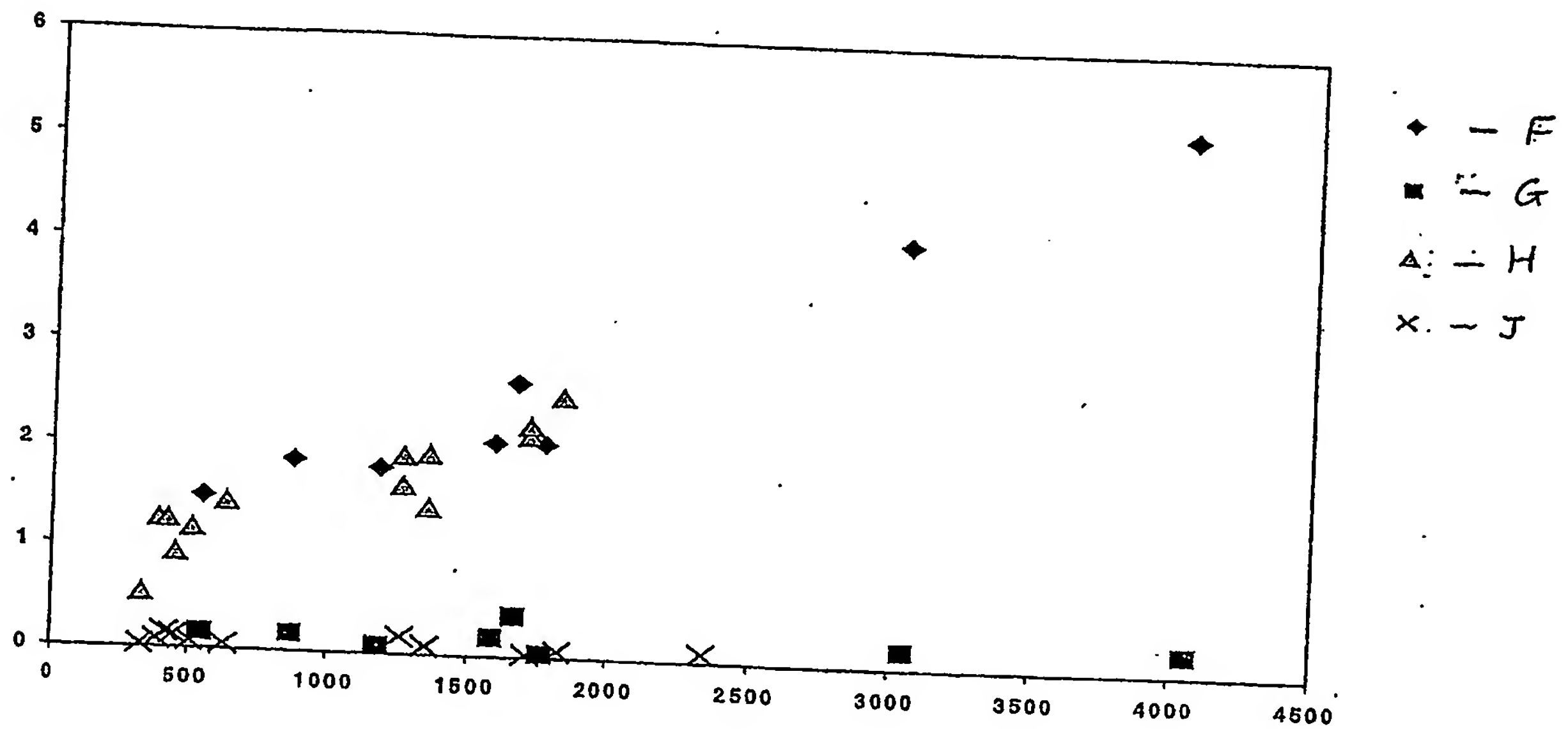


Figure 6

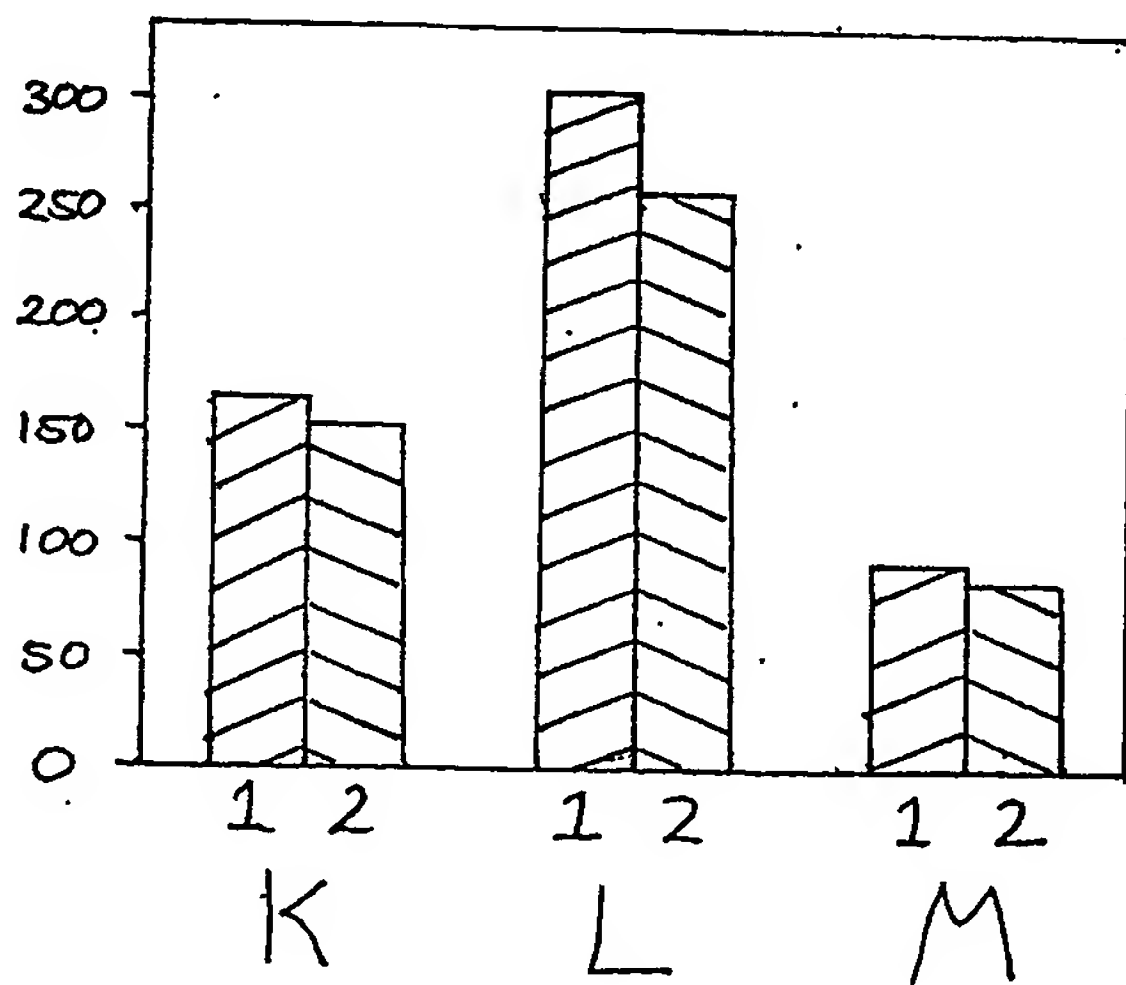
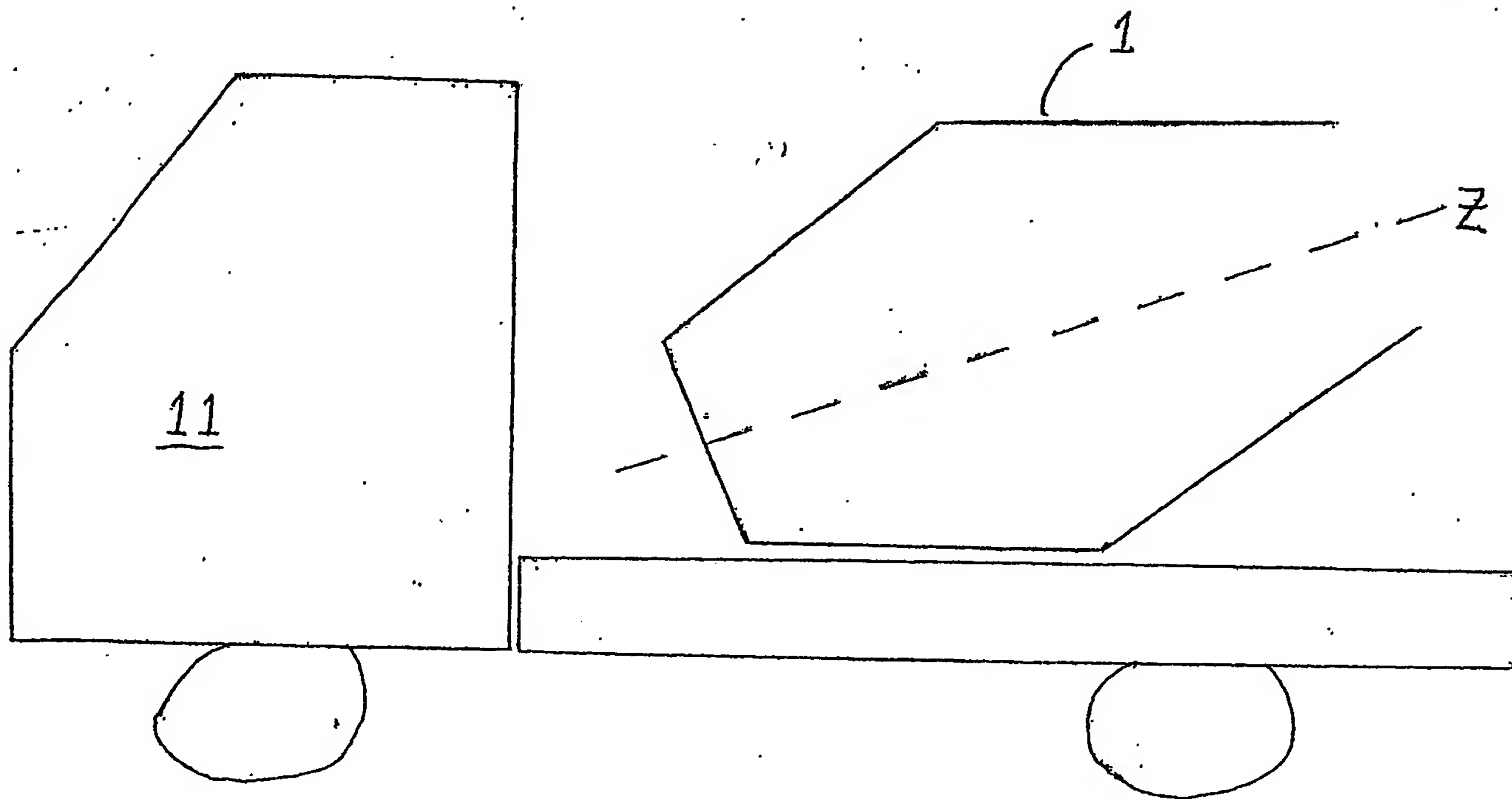


Figure 7



BEST AVAILABLE COPY

THIS PAGE BLANK (USPTO)

9 PCT/EP2004/005968

